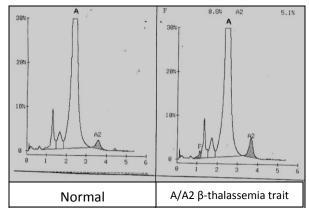
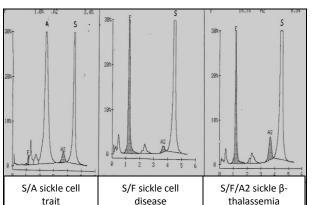
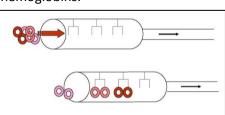
- Principle of electrophoresis: movement of charged molecules from cathode (-) to anode (+) in a gel under the influence of an electrical field.
- The rate of migration of the molecules is dependent on:
 - Charge of the molecules.
 - Size. •
 - Shape.
 - Buffer pH. •
 - Time frame of the procedure.
 - Temperature of the operating system.
- Protein electrophoresis:
 - More complex than DNA electrophoresis • (WHY?)
 - ✓ Different proteins have different charges.
 - \checkmark Proteins vary widely in their shape.
 - Gel medium: polyacrylamide.
 - Separation of proteins occur under:
 - Denaturing conditions: in which proteins are treated with anionic detergents so they will maintain their charge but will lose their normal shape.
 - *Non-denaturing conditions*: in which there will be no pre-treatment of proteins prior to electrophoresis so they will retain both their charge and normal shape.
- Hemoglobin electrophoresis:
 - Aim: to diagnose hemoglobin pattern and any presence of abnormal hemoglobins. •
 - Medium:
 - ✓ Cellulose acetate at alkaline pH (8.9).
 - ✓ Citrate agar at acidic pH (6).
 - Nowadays, it is widely replaced by HPLC. •
- HPLC:
 - Positively charged molecules (salt and hemoglobin) will bind to • negatively charged carboxyl groups (bound to silica).
 - Hemoglobin variants will separate out due to variation in charge.

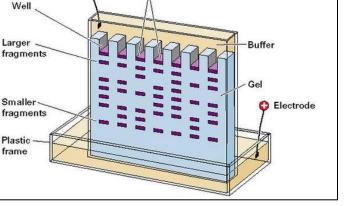
Advantages Disadvantages 1. Automation: less time – more samples analyzed 1. Expensive initial setup. 2. Samples: very small (5µl). 2. Requires periodic optimization and skillful 3. Versatility: simultaneous quantification of HbA, operation. HbA2, HbF, HbS and HbC. 3. HbS co-elute with HbA2 rendering its 4. Sensitivity: detecting δ chain in β -thalassaemia quantification inaccurate. minor. 5. Quantitation.







Positively charged haemoglobins attach to the carboxyl groups at varying strengths



Samples

Electrode

HPLC vs. electrophoresis